

## Wide Range Nematic Mixtures Incorporating 4''-n-Alkyl-4-Cyano-*p*-Terphenyls

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*Summary* The 4''-n-alkyl-4-cyano-*p*-terphenyls permit the formulation of colourless, stable, wide range nematic mixtures of simple eutectic compositions, valuable for electro-optic displays, and the readily prepared 4''-bromo-4-n-heptanoyl-*p*-terphenyl provides a rare example of a direct smectic B to nematic transition.

RECENTLY,<sup>1,2</sup> we reported on the liquid crystalline behaviour of the 4'-n-alkyl- and 4'-n-alkoxy-4-cyano-biphenyls. The n-alkyl compounds are particularly low melting, and a number are nematic (or smectic) at room temperature. As a result of their large positive dielectric anisotropies, the materials perform well<sup>3</sup> in electro-optic displays based on the twisted nematic effect. Furthermore, the compounds are colourless, and chemically and photochemically stable. Using eutectic mixtures,<sup>4</sup> the

nematic ranges may be widened to 0–60 °C making the materials attractive for electro-optic displays.

The desire for even wider range nematic materials necessitates the incorporation of other compounds of much higher nematic thermal stabilities. We have therefore synthesised new colourless 4''-n-alkyl-4-cyano-*p*-terphenyls (Table 1) exhibiting high nematic thermal stabilities, allied with chemical-photochemical stability.

The compounds were examined by microscopy and differential thermal analysis. A number exhibit solid polymorphism and enthalpies for the different transitions were determined.

The importance of these new materials lies in their mixtures with the low melting biphenyls. Eutectic mixtures have low melting points and high N-I temperatures (Table 2).

TABLE 1. Transition temperatures for the compounds

| R   | 4''-R·C <sub>6</sub> H <sub>4</sub> ·C <sub>6</sub> H <sub>4</sub> ·C <sub>6</sub> H <sub>4</sub> ·CN-4 |                                      |
|---|---|--------------------------------------|
|   | K-N or S <sub>E</sub><br>Temp.<br>(t/°C)  | S <sub>A</sub> -N<br>Temp.<br>(t/°C) |
| n-C <sub>8</sub> H <sub>7</sub>               | 182   | —                                    |
| n-C <sub>8</sub> H <sub>11</sub>              | 130   | —                                    |
| n-C <sub>8</sub> H <sub>13</sub>              | 125   | —                                    |
| n-C <sub>8</sub> H <sub>15</sub>              | 134   | (125·5)                              |
| n-C <sub>8</sub> H <sub>17</sub> <sup>a</sup> | 127   | 197                                  |
|   |   | N-I<br>Temp.<br>(t/°C)               |
|   |   | 257·5                                |
|   |   | 239                                  |
|   |   | 228                                  |
|   |   | 222                                  |
|   |   | 216                                  |

<sup>a</sup> Microscopic and miscibility studies have confirmed smectic polymorphism in this compound: S<sub>E</sub>-S<sub>B</sub>, 128 °C and S<sub>B</sub>-S<sub>A</sub>, 133 °C.

K = crystal; S = smectic; N = nematic; I = isotropic liquid. Temperature in parentheses is for a monotropic transition (S<sub>A</sub>-N).

Ternary mixture (2) provides a nematic range equivalent to the best achievable for a *quinary* mixture of biphenyls.<sup>4</sup> Crystallisation occurs only slowly at -15 °C when the mixture is seeded. The low melting mixture (3) has an N-I

TABLE 2. Some predicted eutectic mixtures of 4''-n-pentyl-4-cyano-*p*-terphenyl (5CT)<sup>a</sup> and 4'-substituted (R) 4-cyanobiphenyls and the actual temperature ranges

| Mixture | Composition<br>R/5CT   | Molar % | Predicted              |               | Actual                 |               |
|---------|--|---------|------------------------|---------------|------------------------|---------------|
|         |  |         | K-N<br>Temp.<br>(t/°C) | N-I<br>(t/°C) | K-N<br>Temp.<br>(t/°C) | N-I<br>(t/°C) |
| (1)     | n-C <sub>5</sub> H <sub>11</sub><br>5CT  | 86·7    | 17                     | 64            | 14                     | 66            |
|         |  | 13·3    |                        |               |                        |               |
| (2)     | n-C <sub>5</sub> H <sub>11</sub><br>n-C <sub>7</sub> H <sub>15</sub><br>5CT  | 52·6    | 0·4                    | 63·5          | 0                      | 62·5          |
|         |  | 35·4    |                        |               |                        |               |
|         |  | 12·0    |                        |               |                        |               |
| (3)     | n-C <sub>8</sub> H <sub>7</sub> O<br>n-C <sub>5</sub> H <sub>11</sub> O<br>n-C <sub>7</sub> H <sub>15</sub> O<br>n-C <sub>8</sub> H <sub>17</sub> O<br>5CT | 23·1    | 9                      | 91            | 5                      | 91            |
|         |  | 22·9    |                        |               |                        |               |
|         |  | 18·8    |                        |               |                        |               |
|         |  | 23·9    |                        |               |                        |               |
|         |  | 11·3    |                        |               |                        |               |

<sup>a</sup> Enthalpy of melting ( $\Delta H$ ) for 5CT = 4·06 kcal mol<sup>-1</sup>.

K = crystal; N = nematic; I = isotropic liquid.

<sup>1</sup> G. W. Gray, K. J. Harrison, and J. A. Nash, *Electron. Letters*, 1973, 9, 130.

<sup>2</sup> G. W. Gray, K. J. Harrison, J. A. Nash, J. Constant, D. S. Hulme, J. Kirton, and E. P. Raynes, Proceedings of 166th National A.C.S. Meeting on Ordered Fluids and Liquid Crystals, Chicago, August 1973, eds., R. S. Porter and J. F. Johnson, to be published.

<sup>3</sup> A. Ashford, J. Constant, J. Kirton, and E. P. Raynes, *Electron. Letters*, 1973, 9, 118.

<sup>4</sup> D. S. Hulme, E. P. Raynes, and K. J. Harrison, *J.C.S. Chem. Comm.*, 1974, 98.

<sup>5</sup> D. Demus, M. Klapperstück, R. Rurainki, and D. Marzotko, *Z. phys. Chem. (Leipzig)*, 1971, 246, 385.

temperature 30 °C higher than that attainable using biphenyls alone.<sup>4</sup> As with the biphenyls,<sup>3</sup> the predicted and observed eutectic temperatures (m.p. and N-I) are in good agreement. Furthermore, introducing a terphenyl does not adversely affect the electrical properties of the mixtures or their performance in electro-optic displays.

The 4''-n-alkyl-4-cyano-*p*-terphenyls were synthesised from 4-bromo-*p*-terphenyl by a method<sup>2</sup> analogous to that for the biphenyl analogues. The intermediate 4''-n-alkanoyl-4-bromo-*p*-terphenyls are liquid crystalline (Table 3).

TABLE 3. Transition temperatures for the compounds 4''-R·CO·C<sub>6</sub>H<sub>4</sub>·C<sub>6</sub>H<sub>4</sub>·C<sub>6</sub>H<sub>4</sub>·Br-4

| R                                | K-N<br>or S <sub>E</sub><br>Temp.<br>(t/°C) | S <sub>E</sub> -S <sub>B</sub><br>Temp.<br>(t/°C) | S <sub>B</sub> -S <sub>A</sub><br>Temp.<br>(t/°C) | S <sub>B</sub> or<br>S <sub>A</sub> -N<br>Temp.<br>(t/°C) | N-I<br>Temp.<br>(t/°C) |
|----------------------------------|---|---|---|---|------------------------|
|                                  | C <sub>2</sub> H <sub>5</sub>               | 224   | —   | —   | —                      |
| n-C <sub>4</sub> H <sub>9</sub>  | 204   | —   | —   | —   | 250                    |
| n-C <sub>6</sub> H <sub>11</sub> | 205·5                                       | —   | —   | —   | 241                    |
| n-C <sub>8</sub> H <sub>13</sub> | 178   | 203·5   | —   | 212   | 239                    |
| n-C <sub>7</sub> H <sub>15</sub> | 175   | 204   | 211·5   | 218   | 233·5                  |

K = crystal; S<sub>E</sub> = smectic E; S<sub>B</sub> = smectic B; S<sub>A</sub> = smectic A; N = nematic; I = isotropic liquid.

The n-heptanoyl compound exhibits a direct S<sub>B</sub>-N transition previously reported<sup>5</sup> for only two compounds.

As a result of the usefulness of the new cyano-*p*-terphenyls in mixtures for displays and the interesting smectic polymorphism of the ketones, other homologues of the series are being investigated.

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(Received, 12th March 1974; Com. 285.)